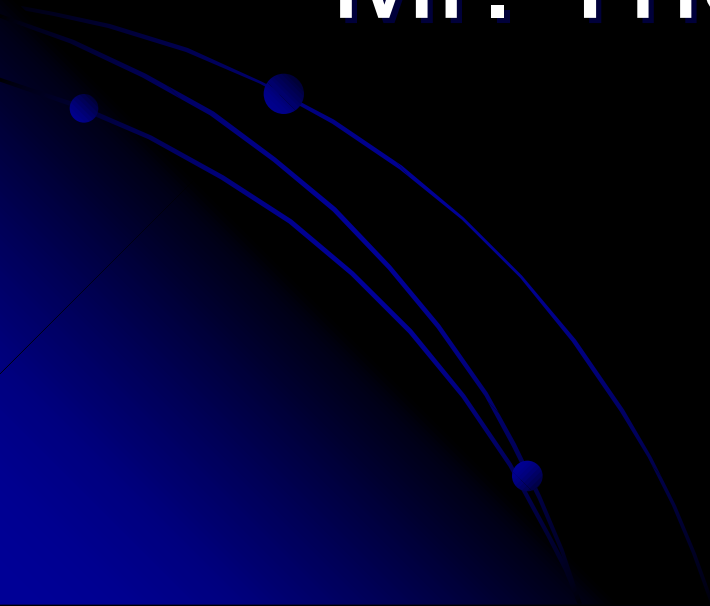


Earth Science

With

Mr. Thomas



Minerals

All rocks & minerals on earth are made of elements.
Minerals have the following characteristics:

- Occurs Naturally
How is a rock different than an mineral?
- It is Solid
- It has a definite chemical composition
Rocks are made of minerals!!
- Its atoms are arranged in an orderly pattern
- It is inorganic (never living)

How Do Minerals Form?

- There are 3 ways:
- From Molten Rock or Magma.
 - The faster it cools, the smaller the crystals!
- From Evaporating Water.
 - Forms Salts!
- From Immense Pressure.
 - Metamorphism – the changing of one rock type to another.

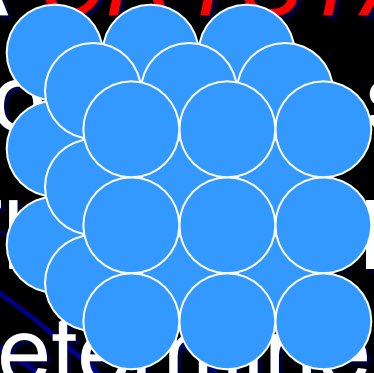
Mineral Structure:

- A mineral's structure is dependent upon the arrangement of the atoms.

Calcite

Quartz

- A **CRYSTAL** structure is a regular, repeating arrangement of atoms.
- The arrangement of atoms determines the shape of a mineral's crystals:



gular geometric surfaces.

ment of the shape of a mineral's crystals



Mineral Structure:

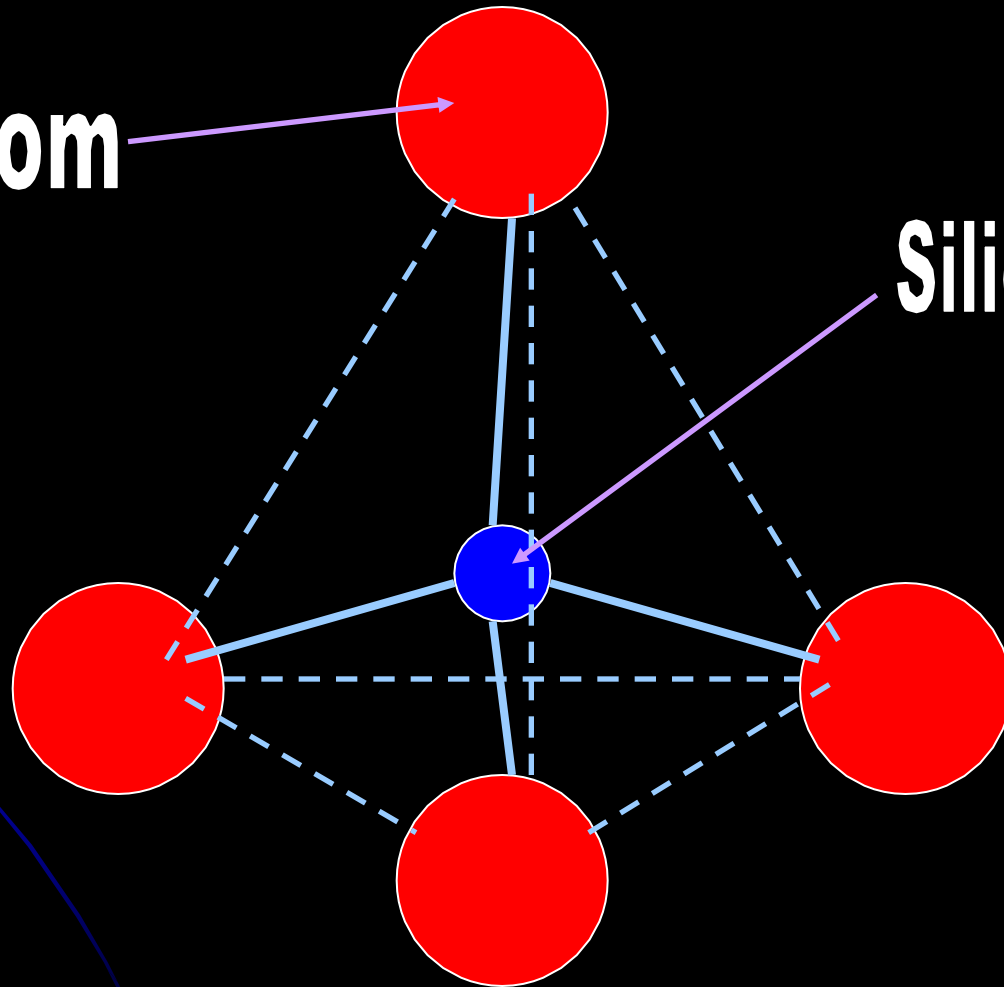
- There are six basic crystal shapes (pg. 99 in text).
- Minerals that contain *Silicon & Oxygen* are called *Silicates*.
- These minerals have a specific crystalline structure:

Silica Tetrahedron

Silica Tetrahedron

Oxygen Atom

Silicon Atom



Mineral Properties

- Cleavage: Tendency for a mineral to break along flat surfaces, on definite planes.
- Fracture: When a mineral breaks into irregular surfaces.
- Hardness: A mineral's resistance to being scratched. Opposite of cleavage.
- Luster: The way a mineral shines in light.

Metallic or Nonmetallic

Mineral Identification

- Identification by Inspection:
 - Mineral Color
 - Luster
 - Cleavage / Fracture

- Identification by Testing:
 - Streak
 - Hardness
 - Acid Test
 - Specific Gravity

Mineral Groups

- Major Silicates:
 - Over 90% of the minerals in earth's crust.
 - Compounds of oxygen, silicon, and a metal.
 - Formed via tetrahedrons.
- Carbonates:
 - Made of negatively charged carbonate ions bonded to positive metal ions. (Covalent Bonds)
- Oxides & Sulfides:
 - Contain significant amounts of IRON combined with either oxygen or sulfur.

Oxygen → Oxide

Sulfur → Sulfide

That's it

